

# Diamond Growth at Michigan State University

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**NSCL / Michigan State University**

**3<sup>rd</sup> NoRHDia Workshop**

**GSI, August 2006**



# Methods of Diamond Growth

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## High-pressure, High-temperature (HPHT)

GE, Sumitomo, De Beers, Gemesis, **not at MSU**

## Chemical Vapor Deposition (CVD)

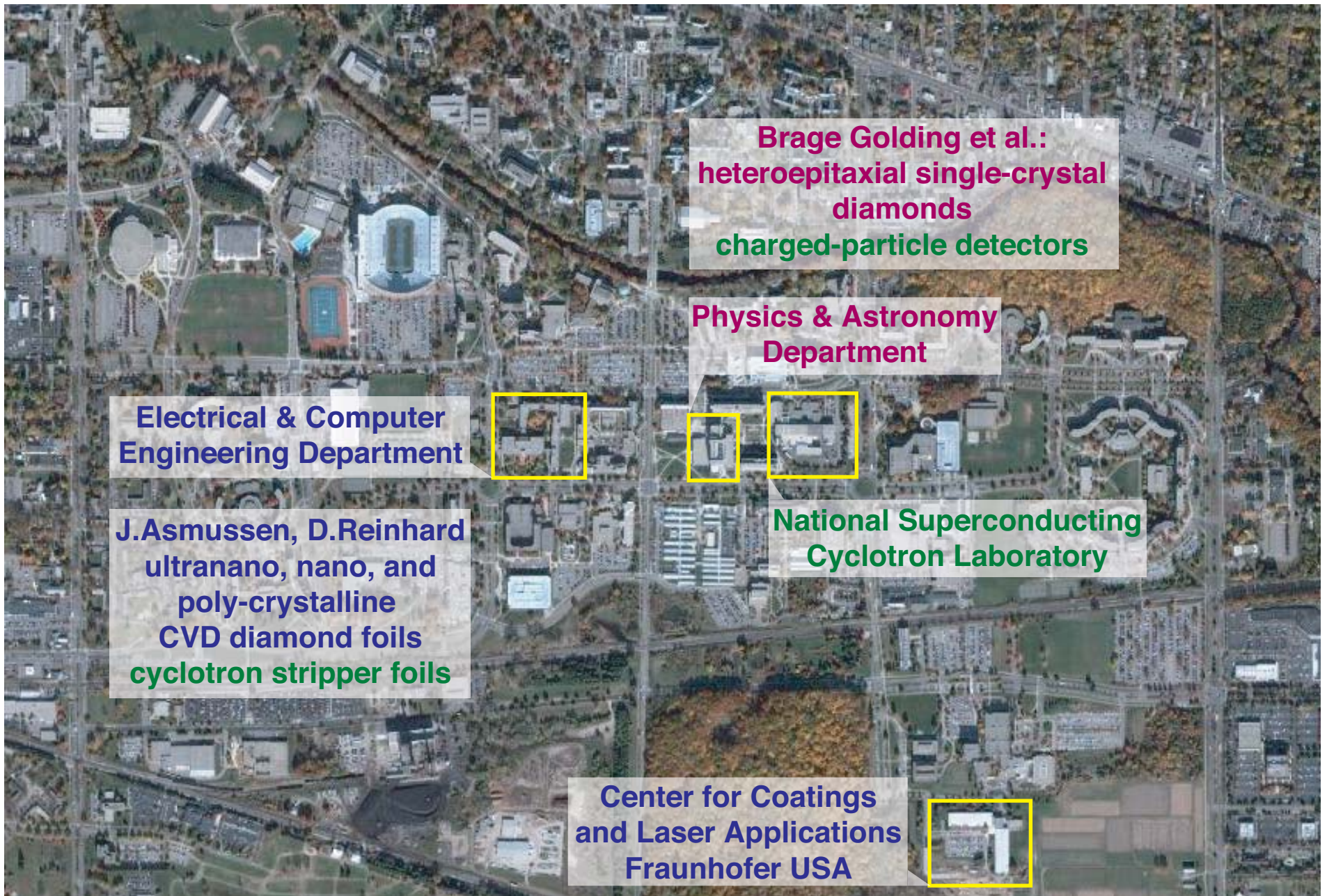
Microwave or hot-filament plasma

Methane and hydrogen, 800 C, 50 Torr

Homoepitaxial (diamond on diamond)

Heteroepitaxial (diamond on non-diamond)

# Diamond Growth at Michigan State University



**Brage Golding et al.:**  
**heteroepitaxial single-crystal diamonds**  
**charged-particle detectors**

**Physics & Astronomy Department**

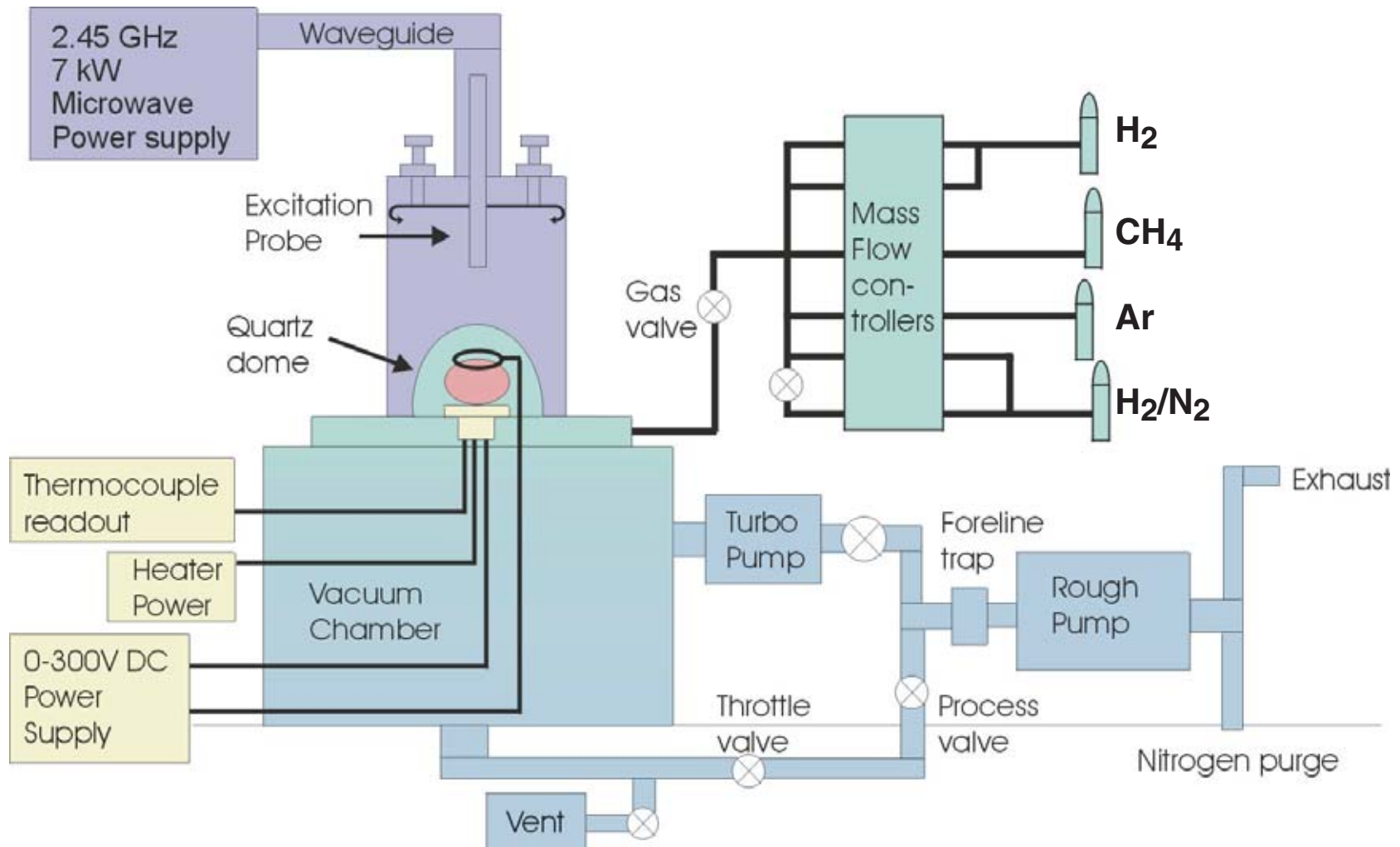
**Electrical & Computer Engineering Department**

**J. Asmussen, D. Reinhard**  
**ultranano, nano, and poly-crystalline CVD diamond foils**  
**cyclotron stripper foils**

**National Superconducting Cyclotron Laboratory**

**Center for Coatings and Laser Applications**  
**Fraunhofer USA**

# Microwave Plasma Chemical Vapor Deposition System



# Growth of diamond on silicon wafers

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## Fabrication and Properties of Ultrananano, Nano, and Polycrystalline Diamond Membranes and Sheets

D. K. Reinhard, J. Asmussen, et al.,  
Michigan State University, Electrical & Computer Engineering Department &  
Fraunhofer Center for Coating and Laser Applications

- **Polycrystalline:**  
grain sizes measured in micrometers
- **Nanocrystalline:**  
Sub-micron grains; grain sizes measured in hundreds of nanometers
- **Ultrananocrystalline:**  
Grain sizes measured in tens of nm and smaller;  
fabricated with Ar/CH<sub>4</sub> chemistry

# Growth of diamond on silicon wafers

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**Nucleation on silicon wafers is facilitated by pre-treatment with diamond powder ( $< 0.25\mu\text{m}$ ). (scratch-seeding stimulates diamond nucleation)**

- **Polycrystalline:**

**Diamond powder is mixed with resist and wafer is spin coated.**

- **Nanocrystalline & Ultrananocrystalline:**

**Diamond dry-polished with powder and cleaned.**

D. K. Reinhard, et al., "Fabrication and properties of ultra-nano, nano, and polycrystalline Diamond Membranes and Sheets", AVS 50th International Symposium, November 2-7, 2003.

# Growth of diamond on silicon wafers

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## Deposition Parameters:

### poly-and nanocrystalline

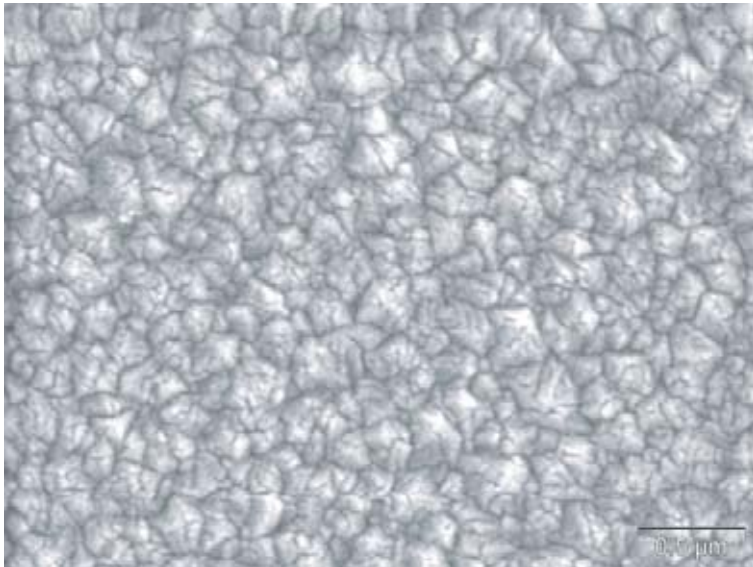
- Pressure:7 –15 torr (poly);  
15 –35 torr(nano)
- Microwave Power: 700 W
- Typical Gas Flow Rates:
  - H2 200 sccm
  - CO2 8 sccm
  - CH4 3 sccm
- Substrate Temperature:
  - 500 - 600°C (poly);
  - 600 - 700°C (nano)

### ultrananocrystalline

- Pressure:120 torr
  - Microwave Power: 1.1 KW,
  - Typical Gas Flow Rates:
    - Ar 100 sccm
    - H2 1 to 4 sccm
    - CH4 1 sccm
  - Substrate Temperature:< 650 C.
- 3” wafer,  
linear growth rate: 1 micron / hour

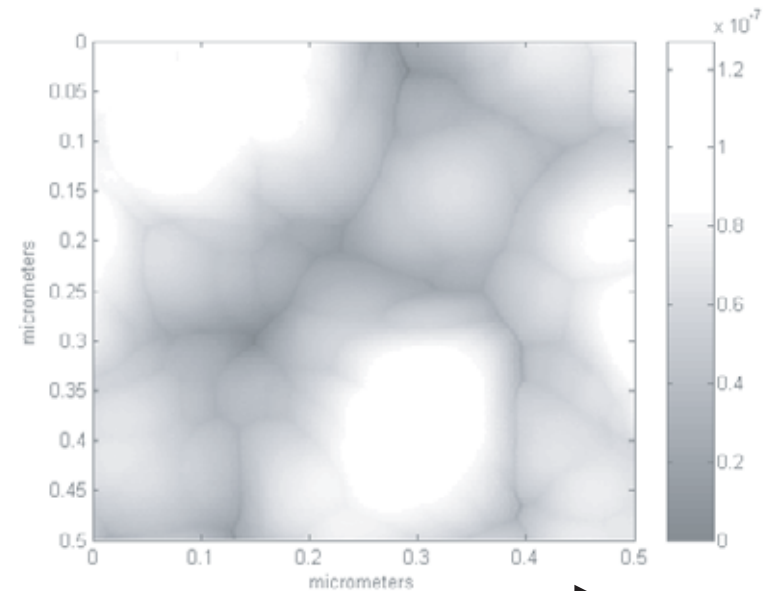
# Nanocrystalline Diamond Film

FB16 SEM



1 μm

AFM



500 nm

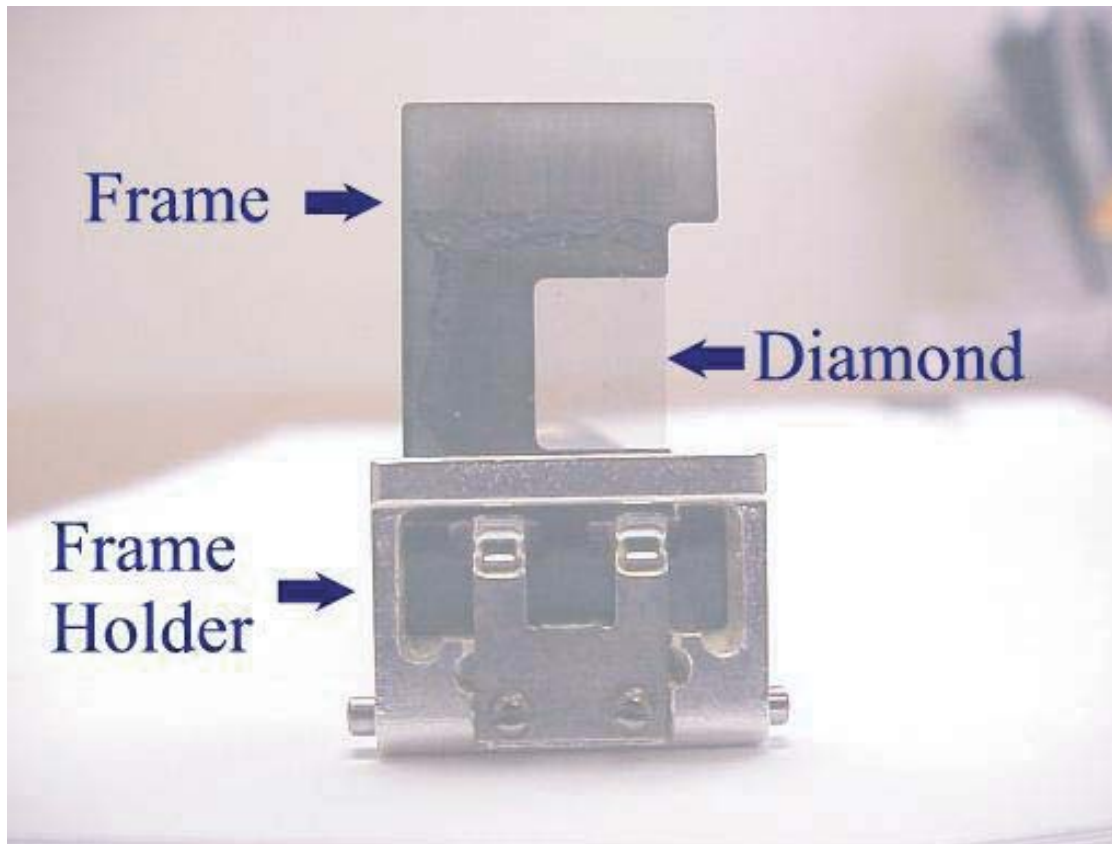
H<sub>2</sub>/CO<sub>2</sub>/CH<sub>4</sub> (sccm): 200/8/3. Power = 0.7 kW, P = 33 torr  
Polished Seeding

D. K. Reinhard, et al., "Fabrication and properties of ultra-nano, nano, and polycrystalline Diamond Membranes and Sheets", AVS 50th International Symposium, November 2-7, 2003.



# Cyclotron Stripper Foils

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**Nanocrystalline diamond foils mounted on 3-sided frame**

**used inside NSCL cyclotron as electron stripper foil for heavy ion beam**

D. K. Reinhard, et al., "Fabrication and properties of ultra-nano, nano, and polycrystalline Diamond Membranes and Sheets", AVS 50th International Symposium, November 2-7, 2003.

# Heteroepitaxial Single Crystal Diamond

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## Growth:

Brage Golding,  
C. Bednarski, M. Behravan, Z. Dai, A.-P. Li, M. Regmi

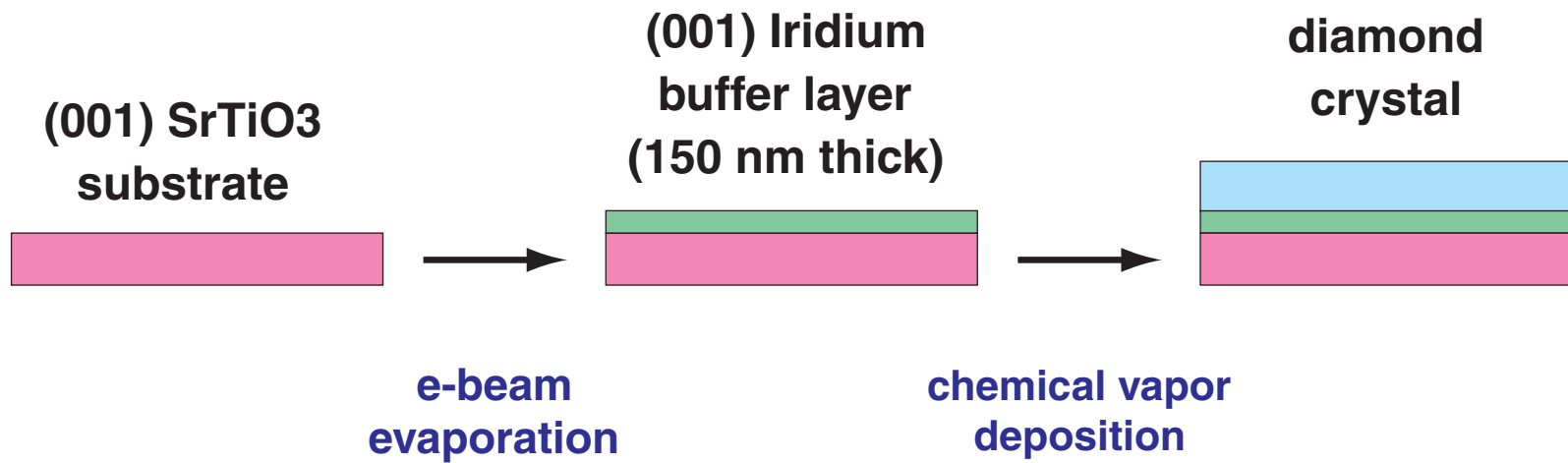
Physics & Astronomy Department, MSU

## Application at NSCL:

diamond detectors for medium-energy heavy-ions,  
timing and tracking applications

# Heteroepitaxial Diamond Growth

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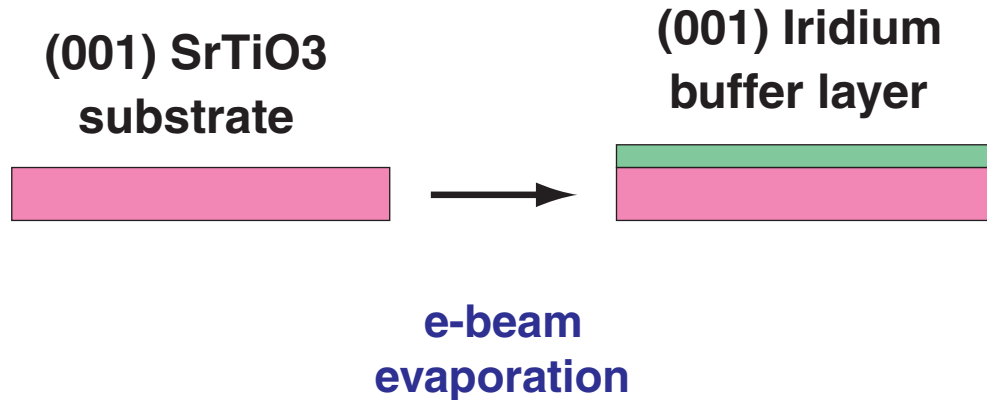


# Material Properties of Substrates

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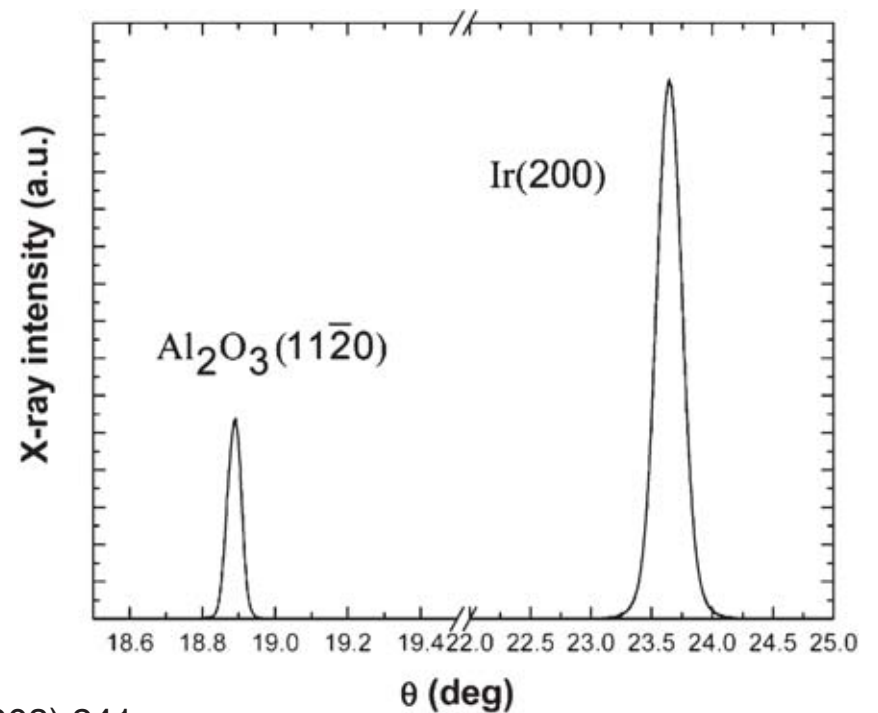
<b>Material</b>	<b>Crystal structure</b>	<b>Lattice constant [nm]</b>	<b>Melting point [C]</b>	<b>Thermal expansivity [10<sup>-6</sup> K<sup>-1</sup>]</b>
<b>Diamond</b>	<b>cubic</b>	<b>0.3567</b>	<b>4027</b>	<b>1.5</b>
<b>Iridium</b>	<b>cubic</b>	<b>0.3839</b>	<b>2484</b>	<b>6.8</b>
<b>Al<sub>2</sub>O<sub>3</sub></b>	<b>hexagonal</b>	<b>a=0.4758 c=1.2991</b>	<b>2040</b>	<b>4.2 5.3</b>
<b>MgO</b>	<b>cubic</b>	<b>0.4216</b>	<b>2800</b>	<b>12.8</b>
<b>SrTiO<sub>3</sub></b>	<b>cubic</b>	<b>0.3905</b>	<b>2080</b>	<b>10.3</b>

# Iridium Growth

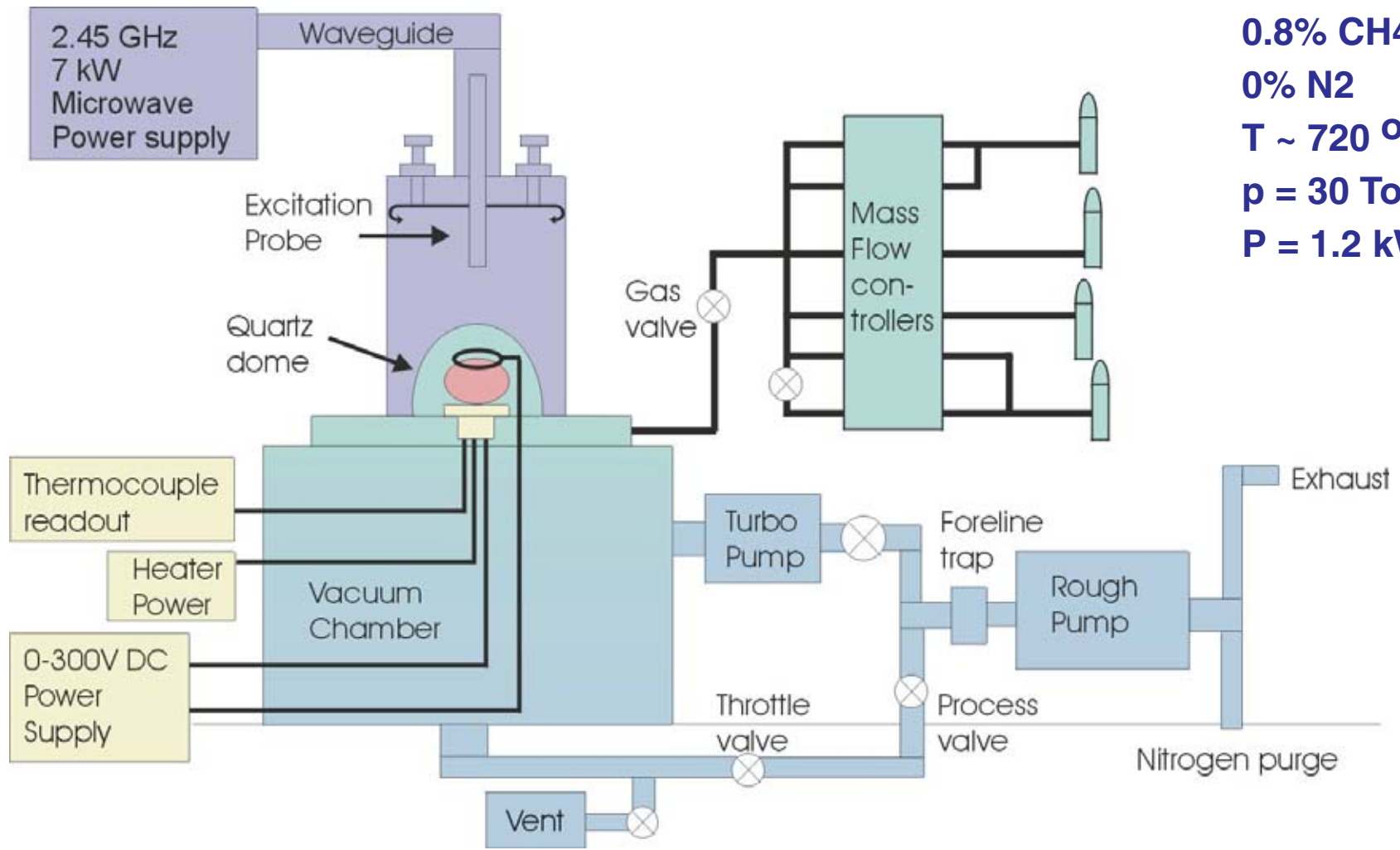


wet chemical cleaning of substrate  
(SrTiO<sub>3</sub> or Al<sub>2</sub>O<sub>3</sub>)  
annealing at 800 C  
e-beam evaporation: 2.4 nm / min  
no post-deposition annealing

X-ray diffraction for 300 nm Ir grown on sapphire



# CVD Process Parameters



**0.8% CH<sub>4</sub>/H<sub>2</sub>**  
**0% N<sub>2</sub>**  
**T ~ 720 °C**  
**p = 30 Torr**  
**P = 1.2 kW**

# Nucleation Process

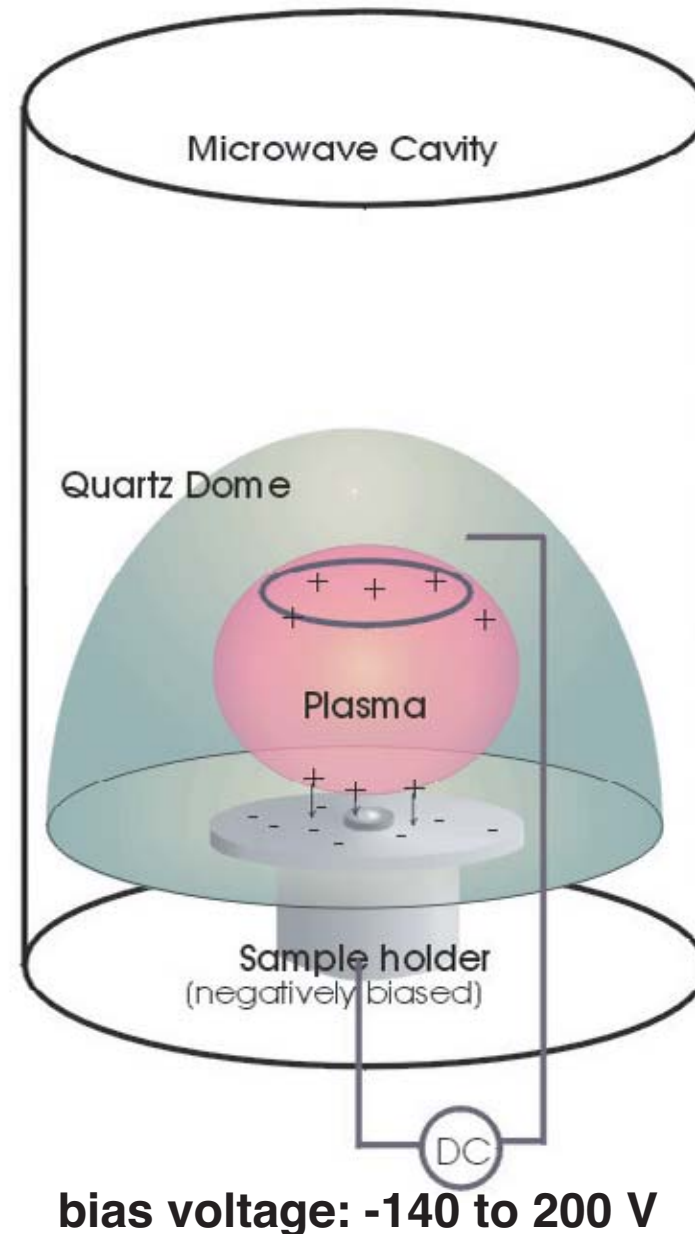
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How do you induce diamond to start growing on a metal?

Schematic of biasing geometry during nucleation process

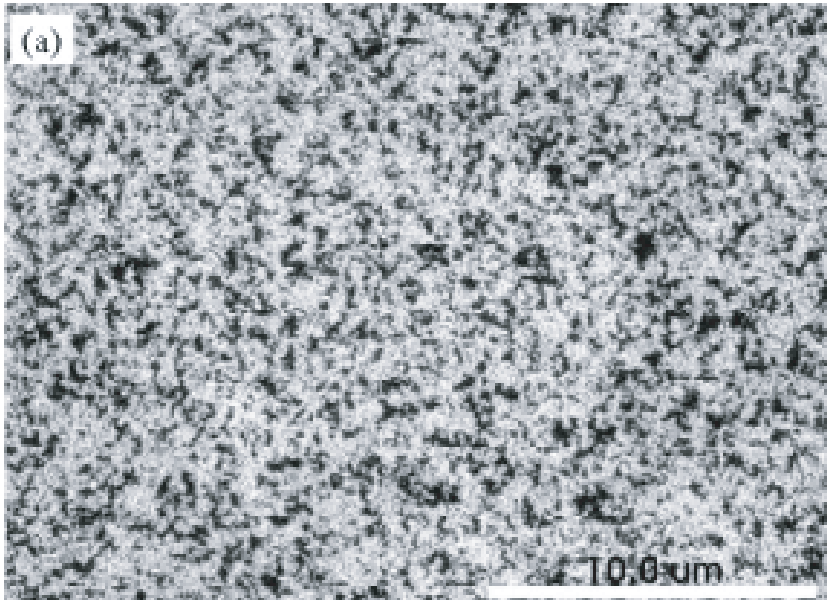
Nucleation occurs with terminating of biasing

With bias: carbon condensate in highly excited non-equilibrium state  
Removal of bias: rapid temperature quench initiating nucleation



# Deposition of Carbon on Ir (001)

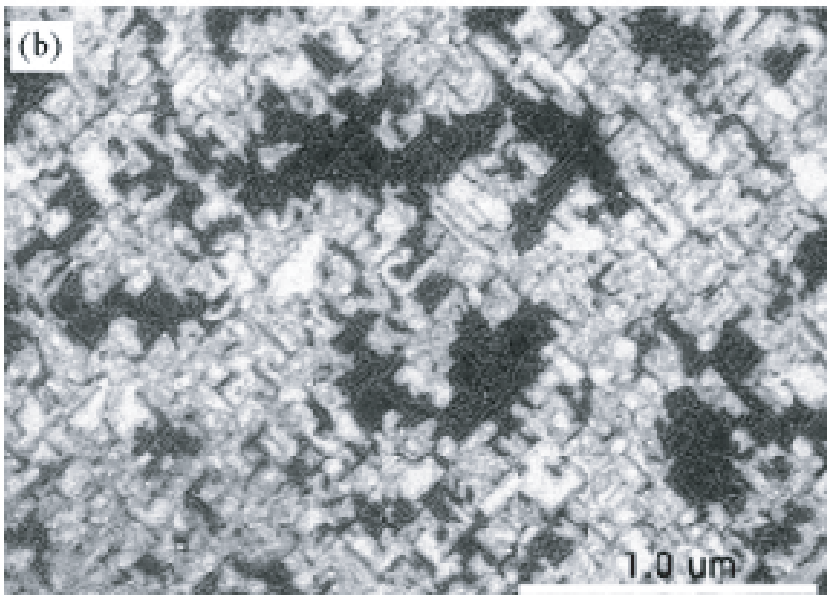
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**Bias period: 20 min**

**bright regions are diamond-like carbon condensate**

**condensate covers 75% of surface**



**diffuse rectangular blocks of condensate**

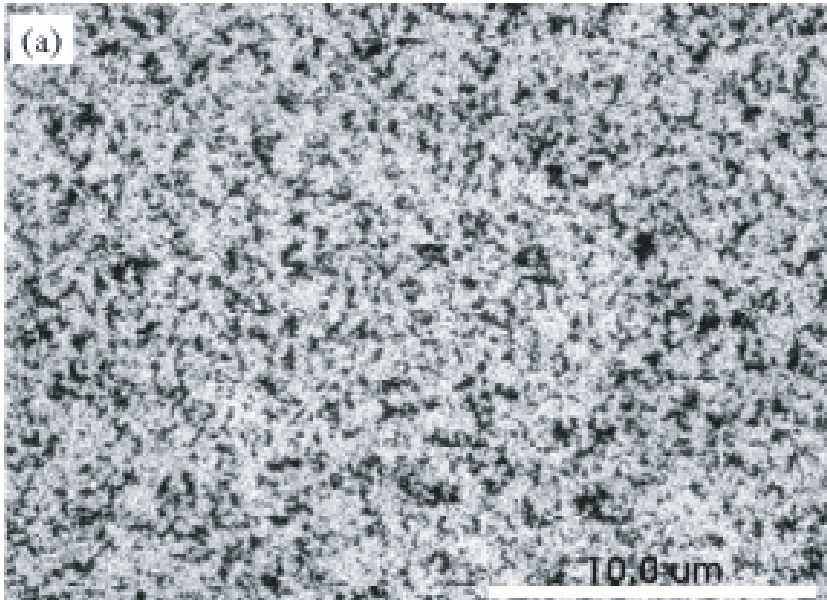
**characteristic linear dimensions: 100-200 nm**

**boundaries parallel to in-plane Ir [110] directions**



# Deposition of Carbon on Ir (001)

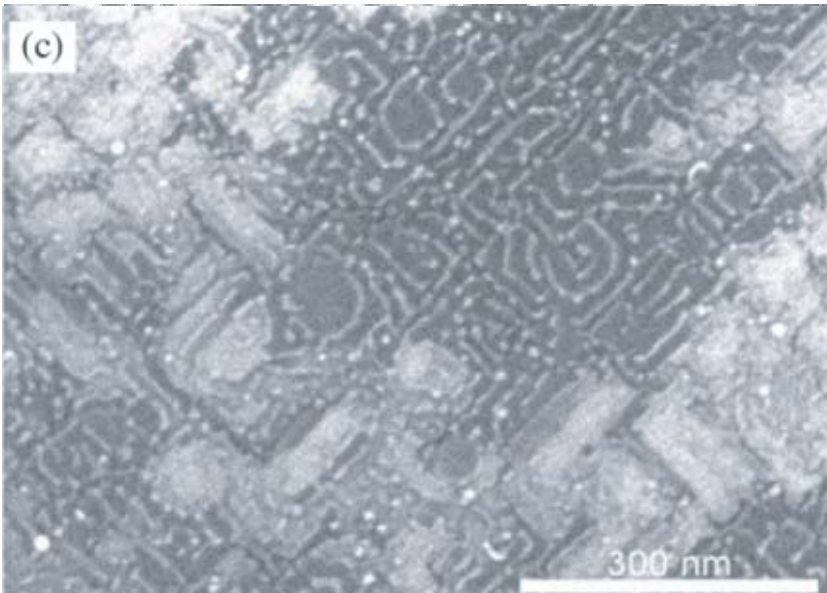
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**Bias period: 20 min**

**bright regions are diamond-like carbon condensate**

**condensate covers 75% of surface**

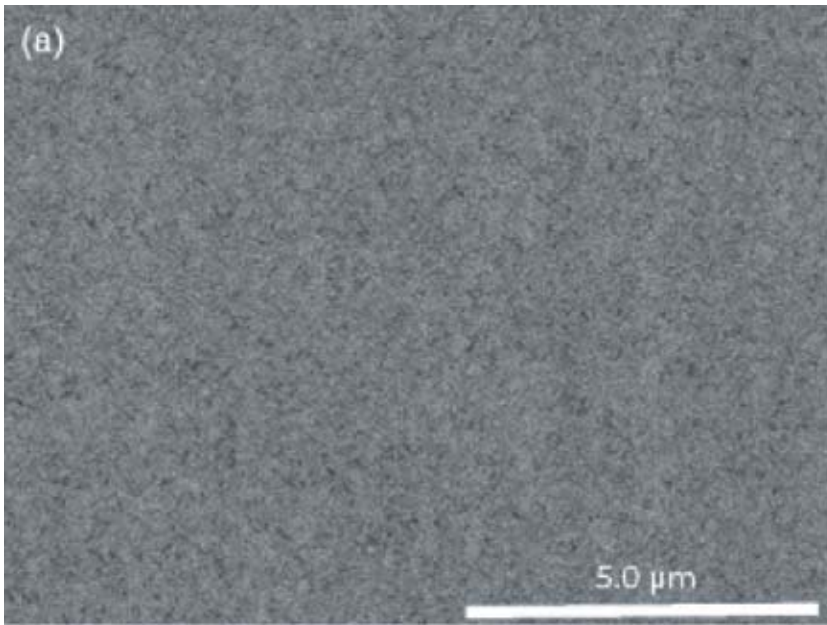


**higher magnification of low-density region:**

**diffuse condensate in the presence of decorated ridges and whorls of Ir surface**

# Deposition of Carbon on Ir (001)

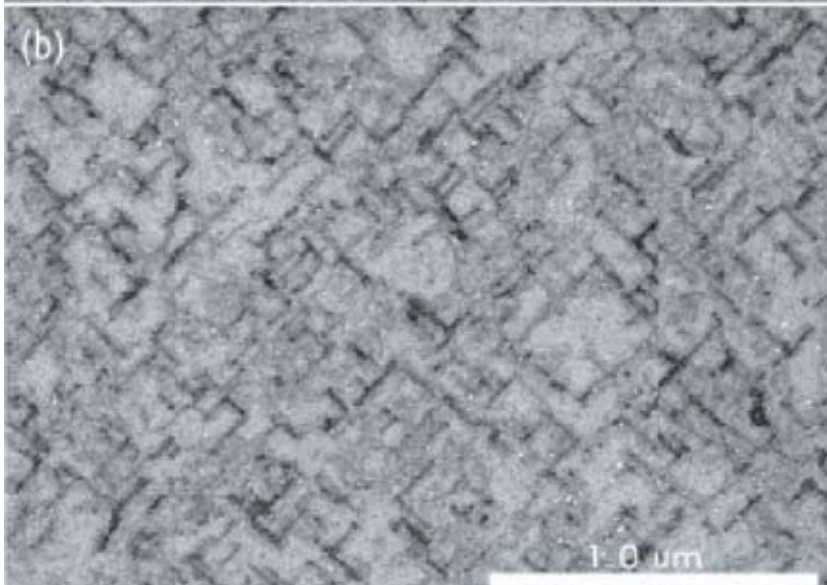
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**Bias period: 60 min**

**bright regions are diamond-like carbon condensate**

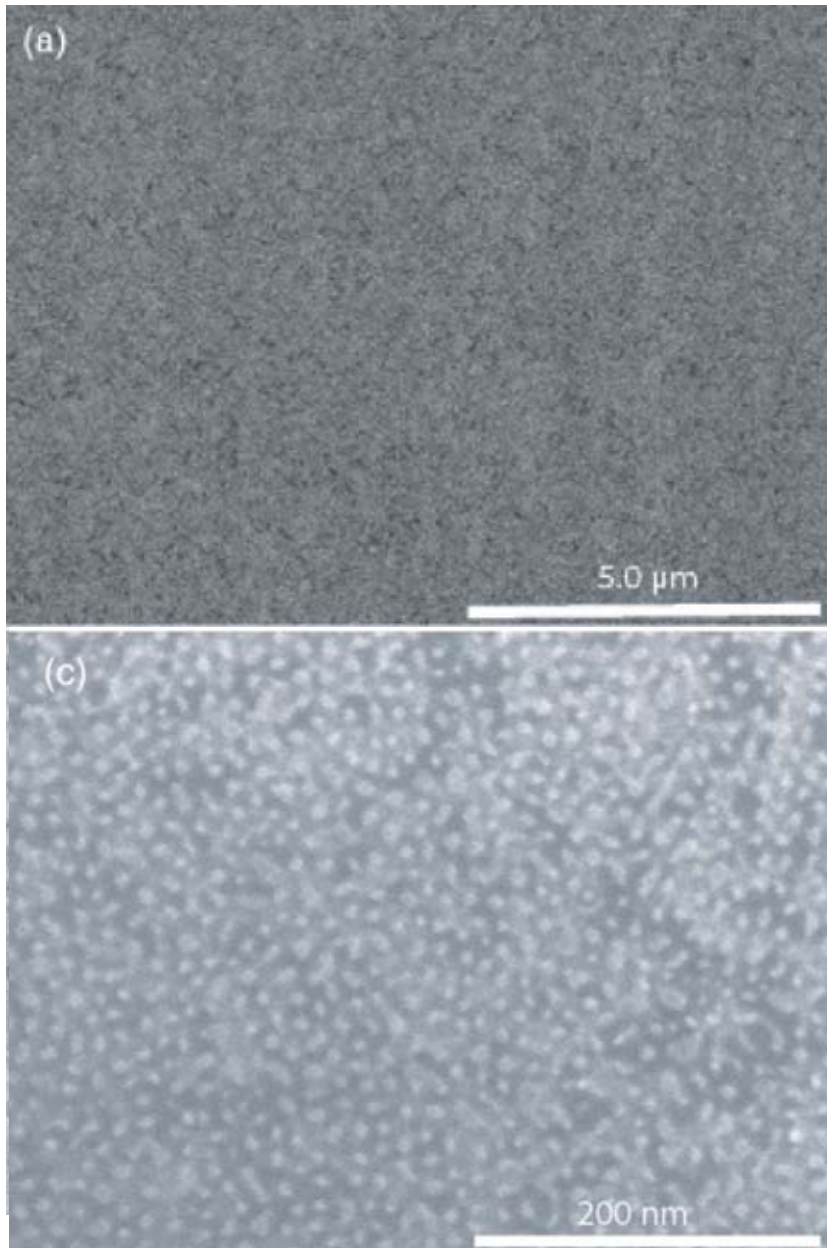
**condensate covers 82% of surface  
increase of thickness of condensate**



**low -density region with <110> alignment**

# Deposition of Carbon on Ir (001)

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**Bias period: 60 min**

**bright regions are diamond-like carbon condensate**

**condensate covers 82% of surface  
increase of thickness of condensate**

**higher magnification reveals diamond nanocrystallites**

**clusters with diameters near 7 nm  
separation ~ twice diameter**

# Early Stages of Diamond Evolution

## 1. Bias

hot plasma discharge  
highly excited carbon  
condensate  
deposition of  $sp^3$  carbon

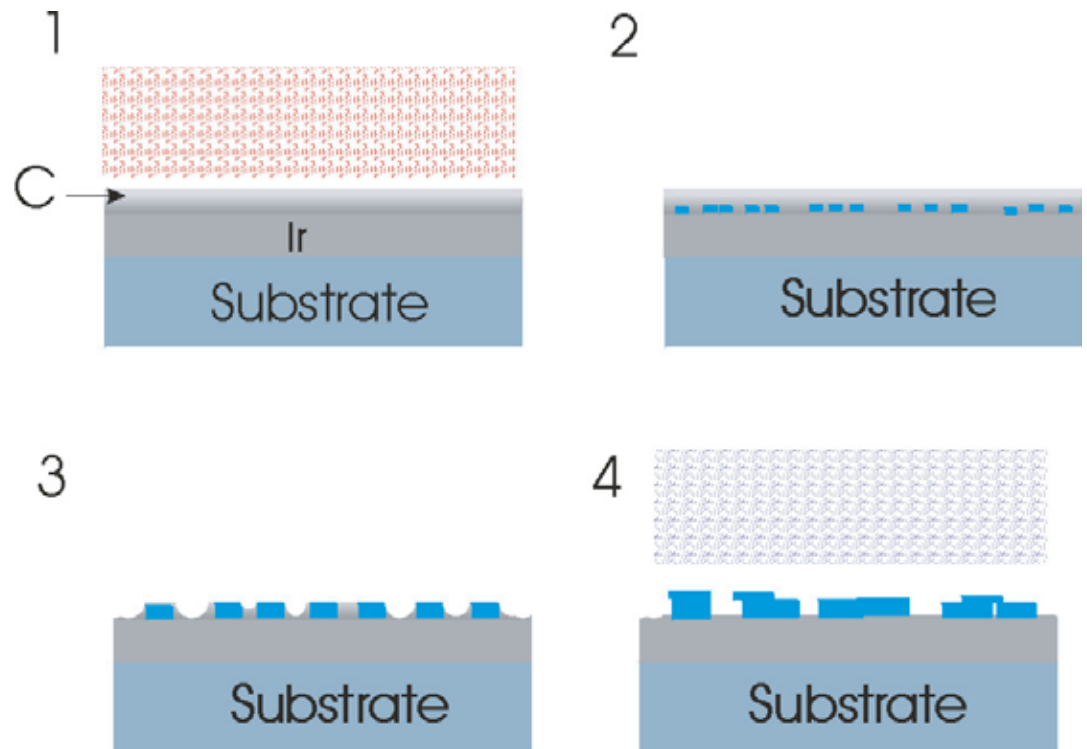
## 2. Quench

condensate cools rapidly  
region of instability  
appearance of nanocrystals

## 3. Growth from matrix

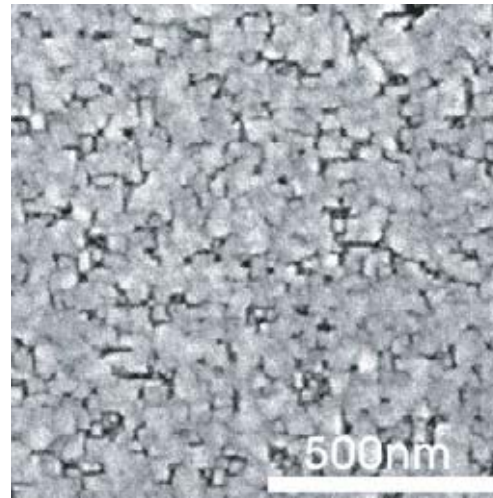
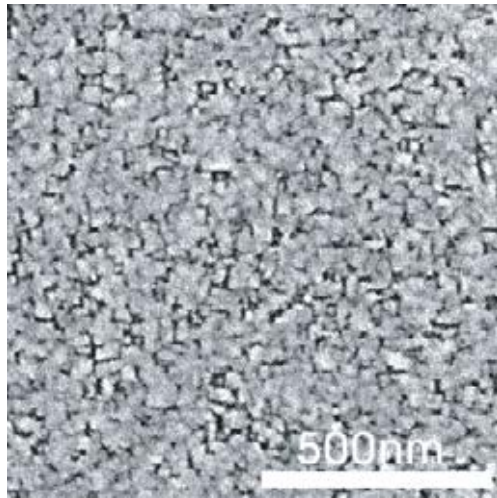
diamond is denser phase  
than precursor  $\rightarrow$   
depletion of condensate

## 4. Coalescence and growth from vapor



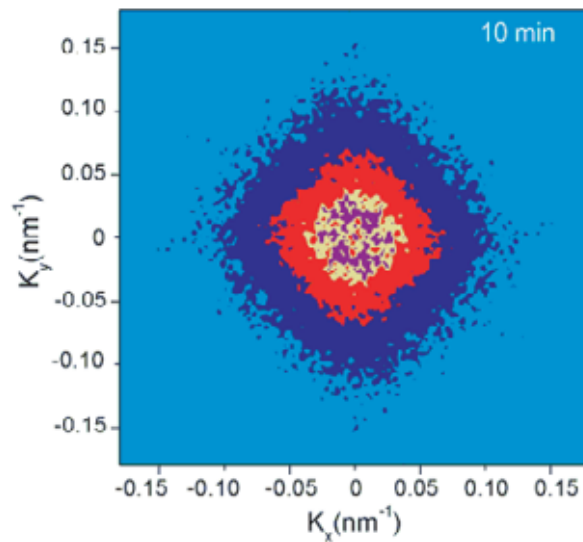
# Growth of Diamond on Ir (001)

## SEM micrographs and 2D-Fourier transforms

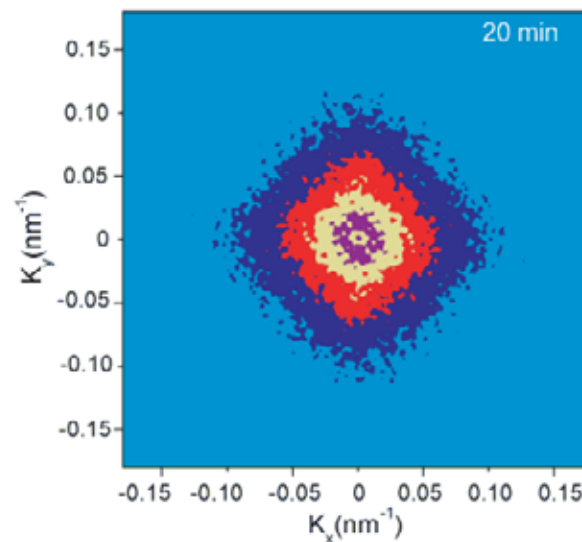


The Ir surface is covered with diamond after 10 min growth.  
(time after nucleation).

Early coalescence due to high nucleation density



**10 min**



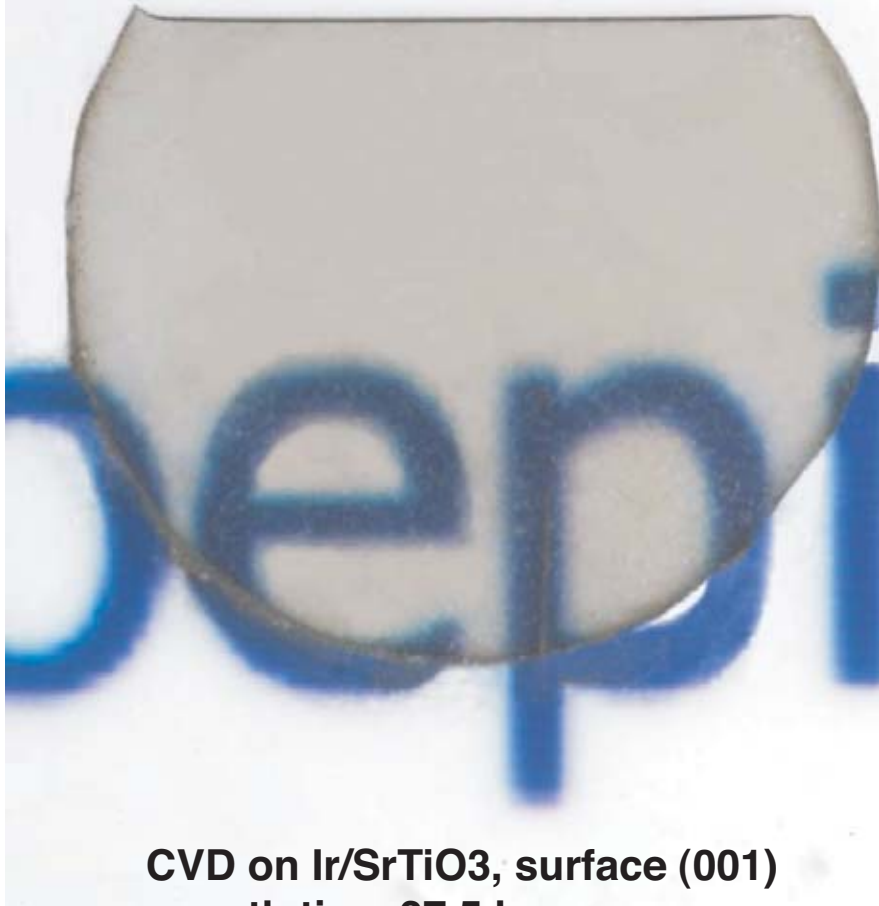
**20 min**

Coalescence occurs after 20 min:  
contraction of  
Fourier transform

# Heteroepitaxial Single-Crystal Diamond

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**3.5 mm diameter, 25  $\mu\text{m}$  thickness  
good optical quality**



**CVD on Ir/SrTiO<sub>3</sub>, surface (001)  
growth time 37.5 hrs**

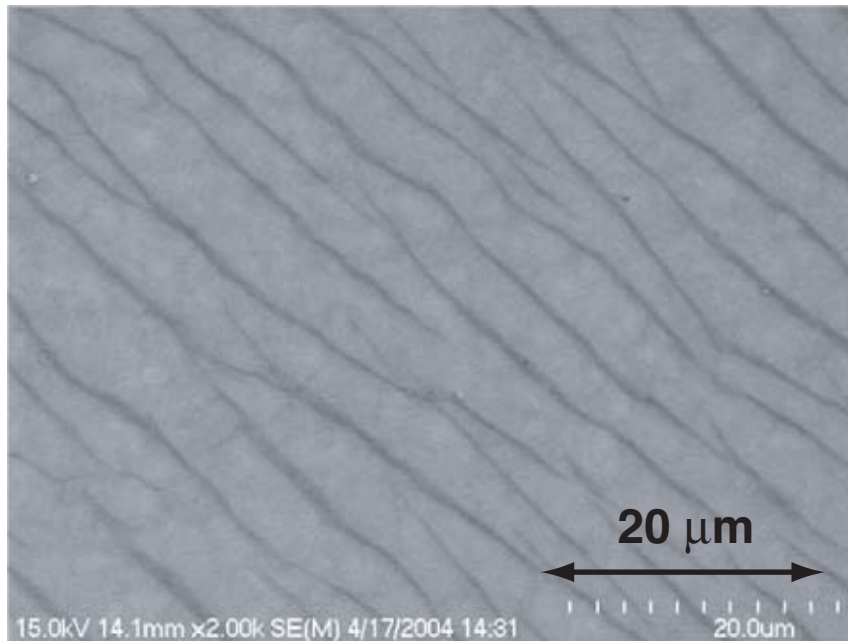
**Commercial  
polycrystalline**



# Heteroepitaxial SC diamond

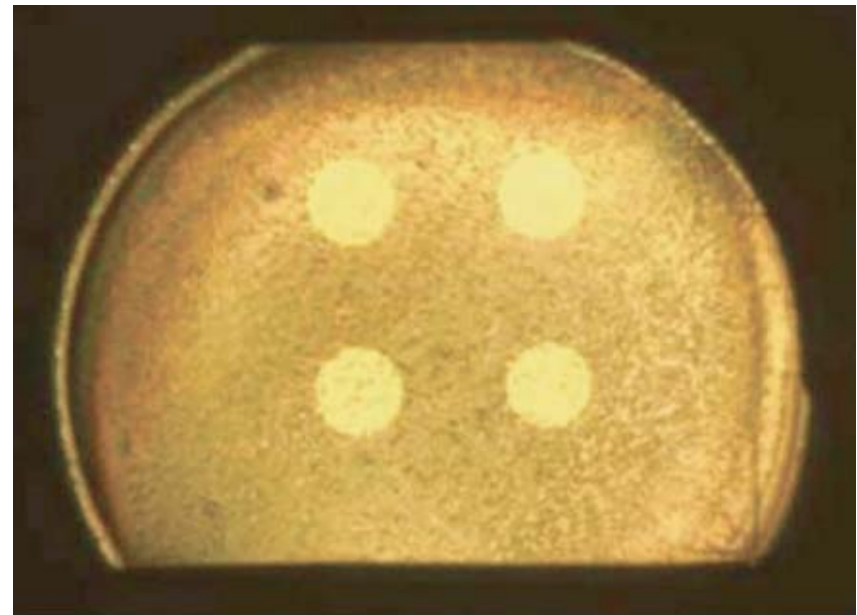
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Scanning electron microscope image  
of the (001) diamond surface



**step bunching, parallel to [110]  
growth by step-flow mode,  
pinning due to imperfections**

optical photograph  
with electrodes



**-> investigated use for  
particle detectors**